

Download File PDF Naming And Writing Formulas For Ionic Compound Chapter 9 Worksheet Answers

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Name _____ KEY _____ Date _____ Period _____
Honors Chem Study

CHAPTER 9
Writing and Naming Ionic Compounds

Write the chemical formula for each of the following chemical names:

1) Mercury (II) nitrate	$Hg(NO_3)_2$	9) Sodium hydroxide	$NaOH$
2) Aluminum iodide	AlI_3	10) Tin (IV) carbide	SnC
3) Potassium fluoride	KF	11) Aluminum sulfate	$(Al_2)_2SO_4$
4) Zinc phosphide	Zn_3P_2	12) Potassium hydroxide	KOH
5) Strontium carbonate	$SrCO_3$	13) Nickel (II) nitrite	$Ni(NO_2)_2$
6) Silver chromate	Ag_2CrO_4	14) Cesium phosphate	Cs_3PO_4
7) Lead (IV) acetate	$Pb(C_2H_3O_2)_4$	15) Magnesium selenide	$MgSe$
8) Lithium chloride	$LiCl$	16) Calcium nitride	Ca_3N_2

Write the chemical name for each of the following chemical formulas:

17) Na_2O	Sodium oxide	25) $CsNO_2$	Cesium nitrite
18) $AgNO_2$	Silver nitrite	26) $FeSO_4$	Iron(II) sulfate
19) $(NH_4)_2CO_3$	Ammonium carbonate	27) $AlPO_4$	Aluminum phosphate
20) $NiCl_2$	Nickel(II) chloride	28) $Mg_3Si_2O_8$	Magnesium silicate
21) HgO	Mercury(II) oxide	29) Sr_3P_2	Strontium phosphide
22) $CaBr_2$	Calcium bromide	30) PbF_2	Lead(II) fluoride (confirm charge)
23) $ZnSO_4$	Zinc sulfate	31) $PbCl_2$	Lead(II) chloride
24) LiH	Lithium hydride	32) $Ni(OH)_2$	Nickel(II) hydroxide

not ionic

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