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so many fake sites. this is the first one which worked! Many thanks

How many moles of NaCl is needed to make a 6 M solution in 50 ml of water?

$$\text{Molarity} = \frac{\text{moles of solute}}{\text{liters of solution}}$$

6 M = moles of solute / 50 ml



NaCl = 58.5 g/mol

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