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so many fake sites. this is the first one which worked! Many thanks

35. The standard reduction potentials for Cu^{2+}/Cu , Zn^{2+}/Zn , $\text{Fe}^{3+}/\text{Fe}^{2+}$, Ag^{+}/Ag and H^{+}/H_2 are +0.34 V, -0.76 V, +0.77 V, +0.80 V and 0.00 V respectively. Choose the strongest reducing agent among the following: (A) Zn (B) Fe (C) Ag (D) Fe²⁺ (E) Cu
36. The number of electrons present in the reduction of one molecule of hydroquinone is: (A) 8 (B) 9 (C) 3 (D) 7 (E) 4
37. For a first order reaction the rate constant is 6.93 min^{-1} . The time taken for 75% conversion in minutes is (A) 3.32 log 2 (B) 2.3 log 3 (C) 2.3 log 2 (D) 3.32 log 4
38. The unit, and $L^{-1} s^{-1}$ is a unit for the rate constant of the reaction having the order (A) 0 (B) 2 (C) 1 (D) 3 (E) 4
39. The half-life of a reaction is inversely proportional to the square of the initial concentration of the reactant. Then the order of the reaction is (A) 0 (B) 1 (C) 2 (D) 3 (E) 0.5
40. The gas which is least absorbed on charcoal (under identical conditions) is (A) HCl (B) O_2 (C) CO_2 (D) NH_3 (E) SO_2
- (Source: IIT-JEE)

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