

Download File PDF Chapter 9 Chemical Names And Formulas Section Review Answers

#Jenny



Finally I get this ebook, thanks for all these I can get now!

#Rio



Cool! I'am really happy

#Markus Jensen



I did not think that this would work, my best friend showed me this website, and it does! I get my most wanted eBook

#Hun Tsu



wtf this great ebook for free?!

#Che Salsa



My friends are so mad that they do not know how I have all the high quality ebook which they do not!

#Diego Butler



so many fake sites. this is the first one which worked! Many thanks

Name: _____ Class: _____ Date: _____

Skills Worksheet
Section 6-1 Concept Review

Section: Simple Ions
Complete each statement below by choosing a term from the following list.
Terms may be used more than once.

10	11	ion	octet
12	13	anion	cation

- An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
- The sodium ion has 11 protons, 10 neutrons, and 12 electrons.
- In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

- An anion is an ion with a negative charge.
- Valency electrons occupy the outermost energy level of an atom.
- The chloride ion and the chlorine atom have different electron configurations.
- A cation is an ion with a positive charge.
- Many stable ions have an electron configuration of a octet.

Answer the following questions in the space provided.

- How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?
noble gases are stable while everything else is missing some

[Download PDF version of :](#)
Chapter 9 Chemical Names And Formulas Section Review Answers