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Name _____ Date _____ Class _____

CHAPTER 12 STUDY GUIDE FOR CONTENT MASTERY

Section 12.2 Forces of Attraction
In your textbook, read about forces of attraction.

Answer the following questions.


1. Ionic, metallic, and covalent bonds are examples of what type of forces? _____
2. Dispersion forces, dipole-dipole forces, and hydrogen bonds are examples of what type of forces? _____
3. Describe dispersion forces.

4. Dispersion forces are greatest between what type of molecules?

5. Describe a permanent dipole.

6. Describe dipole-dipole forces.

7. Describe a hydrogen bond.

8. Identify each of the diagrams below as illustrating dipole-dipole forces, dispersion forces, or hydrogen bonds.

a. _____ b. _____ c. _____
9. Rank dipole-dipole forces, dispersion forces, and hydrogen bonds in order of increasing strength.

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